# Cyclopropanation of N-Substituted 3-Aryl-2-cyanoprop-2enamides and Derivatives of 5,5-Dimethyl-2-oxo-2,5-dihydrofuran-3-carboxylic Acid and 2-Oxochromene-3-carboxylic Acid with Bromine-Containing Zinc Enolates

V. V. Shchepin<sup>a</sup>, P. S. Silaichev<sup>a</sup>, Yu. G. Stepanyan<sup>a</sup>, M. M. Kalyuzhnyi<sup>a</sup>, N. Yu. Russkikh<sup>a</sup>, and M. I. Kodess<sup>b</sup>

<sup>a</sup> Perm State University, ul. Bukireva 15, Perm, 614990 Russia

Received May 4, 2005

**Abstract**—Zinc enolates derived from 1-aryl-2,2-dibromoalkanones react with N-substituted 3-aryl-2-cyano-prop-2-enamides and 5,5-dimethyl-2-oxo-2,5-dihydrofuran-3-carboxylic and 2-oxochromene-3-carboxylic acid derivatives to give, respectively, N-substituted 2-alkyl-3-aryl-2-aroyl-1-cyanocyclopropane-1-carboxamides, 6-(4-bromobenzoyl)-4,4,6-trimethyl-2-oxo-3-oxabicyclo[3.1.0]hexane-1-carboxylic acid ethyl ester and morpholide, and 1-alkyl-1-aroyl-2-oxo-1a,7b-dihydrocyclopropa[c]chromene-1a-carboxylic acids as a single geometric isomer. Treatment of 1-alkyl-1-aroyl-2-oxo-1a,7b-dihydrocyclopropa[c]chromene-1a-carboxylic acids with carboxylic acid anhydrides leads to the formation of the corresponding 9c-alkyl-1-aryl-3,4-dioxo-9b,9c-dihydro-2,5-dioxacyclopenta[2,3]cyclopropa[1,2-a]naphthalen-1-yl carboxylates.

DOI: 10.1134/S1070428006070086

We previously showed that 1-aryl-2,2-dibromobutan-1-ones react with zinc and dialkyl 2-arylmethylidene- or 2-isobutylidenemalonates to give the corresponding cyclopropane derivatives having one aroyl and two ester moieties [1, 2]. With the goal of extending the synthetic potential of this reaction, in the present work we examined reactions of N-substituted 3-aryl-2-cyanoprop-2-enamides with bromine-containing zinc enolates derived from 1-aryl-2,2-dibromoalkanones.

Zinc enolates IIa, IIc, IId, IIf, IIh, and IIi preliminarily prepared from dibromo ketones Ia, Ic, Id, If, Ih, and Ii, respectively, and zinc add at the double bond of electrophilic substrates IIIa—IIId to give intermediates IVa—IVk which undergo intramolecular ring closure to cyclopropane derivatives Va—Vk. Hydrolysis of the latter gives N-substituted 2-alkyl-3-aryl-2-aroyl-1-cyanocyclopropane-1-carboxamides VIa—VIk (Scheme 1). The structure of compounds VIa—VIk was proved by the analytical data and IR and <sup>1</sup>H and <sup>13</sup>C NMR spectra. The IR spectra of VIa—VIk contain absorption bands due to stretching vibrations of the ketone and amide carbonyl groups (1665–1695 cm<sup>-1</sup>)

and NH bond (3325–3405 cm<sup>-1</sup>). In the <sup>1</sup>H NMR spectra we observed signals from protons in the methyl group ( $\delta$  1.50–1.68 ppm, s, R = CH<sub>3</sub>;  $\delta$  0.65–0.82 ppm, t, R = C<sub>2</sub>H<sub>5</sub>) and a singlet from the 3-H proton at  $\delta$  3.51–3.80 ppm. The presence of only one set of signals indicates formation of cyclopropane derivatives **VIa–VIk** as a single stereoisomer.

The steric configuration of compounds VId and VIe was determined by <sup>1</sup>H and <sup>13</sup>C NMR spectroscopy. Their <sup>13</sup>C NMR spectra were consistent with the assumed structure (see Experimental). The carbon signals were assigned on the basis of the HSQC and HMBC two-dimensional correlation spectra. The signals at  $\delta_C$  30.97, 40.73, and 46.23 ppm (VId) and at  $\delta_{\rm C}$  30.50, 40.89, and 52.59 ppm (VIe) were assigned to the C<sup>1</sup>, C<sup>3</sup>, and C<sup>2</sup> nuclei, respectively; these signals are typical of cyclopropane fragment. The configuration of substituents at the cyclopropane fragment in *N*-cyclohexyl-3-(3-bromophenyl)-2-(4-chlorobenzoyl)-1-cyano-2-methylcyclopropan-1-carboxamide (VId) was proved by comparing the  ${}^{3}J_{CH}$  vicinal coupling constants between 3-H and 2-CH<sub>3</sub> with analogous values found for cyclopropa[c]chromenes. We previ-

<sup>&</sup>lt;sup>b</sup> Postovskii Institute of Organic Synthesis, Ural Division, Russian Academy of Sciences, Yekaterinburg, Russia

I, II, R = Me, Ar = Ph (a),  $4\text{-MeC}_6H_4$  (b),  $4\text{-FC}_6H_4$  (c),  $4\text{-ClC}_6H_4$  (d),  $4\text{-BrC}_6H_4$  (e); R = Et, Ar = Ph (f),  $4\text{-MeC}_6H_4$  (g),  $4\text{-FC}_6H_4$  (h),  $4\text{-ClC}_6H_4$  (i); III, Ar' = Ph, R' =  $3\text{-MeC}_6H_4$  (a), PhCH<sub>2</sub> (b),  $2\text{-MeOC}_6H_4$  (c); Ar' =  $3\text{-BrC}_6H_4$ , R' =  $C_6H_{11}$  (d); IV–VI, Ar' = Ph, R' = PhCH<sub>2</sub>, Ar =  $4\text{-FC}_6H_4$ , R = Me (a), Et (b); Ar' =  $3\text{-BrC}_6H_4$ , R' =  $cyclo\text{-C}_6H_{11}$ , R = Me, Ar = Ph (c),  $4\text{-ClC}_6H_4$  (d); Ar' =  $3\text{-BrC}_6H_4$ , R' =  $cyclo\text{-C}_6H_{11}$ , R = Et, Ar =  $4\text{-ClC}_6H_4$  (e); Ar' = Ph, R' =  $3\text{-MeC}_6H_4$ , R = Me, Ar = Ph (f),  $4\text{-ClC}_6H_4$  (g); Ar' = Ph, R' =  $3\text{-MeC}_6H_4$ , R = Me, Ar = Ph (j),  $4\text{-ClC}_6H_4$  (k).

ously synthesized cyclopropanated chromenes with E and Z arrangement of the corresponding hydrogen atom and methyl group, and their configuration was established using NOE techniques, both two-dimensional (2D ROESY) and one-dimensional difference NOE experiments [3, 4]. The coupling constant  ${}^{3}J_{CH}$  in 1-benzoyl-6-bromo-1-methyl-1a-morpholinocarbonyl-1a,7b-dihydrocyclopropa[c]chromen-2-one, in which the H atom and methyl group are located at the opposite sides of the cyclopropane ring (E isomer), was equal to 1.7 Hz, while N-benzyl-1-benzoyl-6-bromo-1methyl-2-oxo-1a,7b-dihydrocyclopropa[c]chromene-1a-carboxamide with Z arrangement of the corresponding moieties was characterized by a  ${}^{3}J_{CH}$  value of 5.3 Hz. The coupling constant  ${}^3J_{\rm CH}$  measured for compound VId is 5.6 Hz, i.e., the 3-H atom and 2-methyl

group in molecule **VId** are located at the same side of the cyclopropane ring (Z isomer). Likewise, similar coupling constants  ${}^3J_{\text{CH}}$  between the same proton and amide carbonyl carbon atom suggest their cis orientation with respect to each other, as in cyclopropa[c]-chromenes.

Thus compound **VId** is a diastereoisomer in which the 3-H proton, 2-methyl group, and 1-carboxamide group reside at the same side of the cyclopropane ring. The 2-CH<sub>2</sub> carbon signal in the  $^{13}$ C NMR spectrum of compound **VIe** appeared as a complex multiplet, and we failed to reliably determine the corresponding coupling constant  $^3J_{\text{CH}}$ . On the other hand, comparison of the chemical shifts of carbon atoms in **VIg** and **VId** indicated that these compounds have the same stereoconfiguration, for  $^{13}$ C chemical shifts are known to be

#### Scheme 2. Br ArCO. ArCO. ZnBr COOEt COOEt ,,,COOEt lle Me Ме –ZnBr₂ Мe Me Me VII IX ΧI Br<sub>.</sub>Me BrZn lle H -ZnBr<sub>2</sub> Ме Ме Me VIII Χ ΧII $Ar = 4-BrC_6H_4$ .

highly sensitive to steric factor. On the basis of the similarity of <sup>1</sup>H NMR spectral patterns of compounds **VId** and **VIe**, on the one hand, and cyclopropane derivatives **VIa**–**VIc** and **VIf**–**VIk**, on the other, the latter were assigned the same configuration.

We previously studied reactions of ethyl 5,5-dimethyl-2,5-dihydro-2-oxofuran-3-carboxylate with bromine-containing zinc enolates derived from 1-aryl-2,2-dibromoalkanones and obtained the corresponding cyclopropanation products, ethyl 6-aroyl-6-ethyl-4,4dimethyl-2-oxo-3-oxabicyclo[3.1.0]hexane-1-carboxylates as a single stereoisomer [5]. However, their structure, specifically the configuration of substituents on C<sup>6</sup>, remained unclear. To elucidate this problem, in the present work we synthesized 6-(4-brombenzoyl)-4,4,6-trimethyl-2-oxo-3-oxabicyclo[3.1.0]hexane-1carboxylic acid ethyl ester XI and morpholide XII. For this purpose, compounds VII and VIII were treated with zinc enolate IIe in diethyl ether-ethyl acetate. Attack by the organozinc reagent on the C<sup>4</sup> atom in VII or VIII gave adduct IX or X, and spontaneous cyclization of the latter resulted in the formation of bicyclic product XI or XII, respectively (Scheme 2). Compound XI showed in the IR spectrum absorption

bands typical of aroyl, ester, and lactone carbonyl groups (1680, 1720, and 1770 cm<sup>-1</sup>, respectively), and the IR spectrum of morpholide **XII** contained absorption bands due to aroyl and amide carbonyl groups (1665 cm<sup>-1</sup>) and lactone carbonyl (1750 cm<sup>-1</sup>). Signals from protons of the methyl group on  $C^6$  and 5-H appeared in the <sup>1</sup>H NMR spectra at  $\delta$  1.59 and 2.59 ppm for ethyl ester **XI** and at  $\delta$  1.44 and 2.80 ppm for morpholide **XII**, respectively.

The  $^{13}$ C NMR spectra of **XI** and **XII** (see Experimental) were fully consistent with the assumed structures. The  $^{13}$ C signals were assigned using HSQC and HMBC two-dimensional correlation techniques. The cyclopropane fragment in molecules **XI** and **XII** is characterized by resonance signals at  $\delta_{\rm C}$  42.61, 42.84, 44.76 (**XI**) and 42.91, 43.61, 46.91 ppm (**XII**). The steric configuration of compounds **XI** and **XII** was determined as described above for **VId** and **VIe**. The coupling constants  $^3J_{\rm CH}$  are 5.2 (**XI**) and 5.4 Hz (**XII**), indicating that the 5-H atom and 6-CH<sub>3</sub> group are located at the same side of the cyclopropane ring plane.

Zinc enolates **IIb**, **IId**–**IIg** obtained from dibromo ketones **Ib** and **Id**–**Ig**, respectively, were brought into reaction with 2-oxochromene-3-carboxylic acid **XIII** 

## Scheme 3.

R = Me,  $Ar = 4-MeC_6H_4$  (a),  $4-ClC_6H_4$  (b),  $4-BrC_6H_4$  (c); R = Et, Ar = Ph (d),  $4-MeC_6H_4$  (e).

#### Scheme 4.

$$XVb$$

$$MeOH$$

$$C_6H_4Cl-p$$

$$Ild$$

$$-ZnBr_2$$

$$XVII$$

$$XVII$$

(Scheme 3). The reaction involved 2 equiv of zinc enolate II which added in succession at the carboxy group and C<sup>4</sup> of molecule XIII to give intermediates XIVa-XIVe. The subsequent intramolecular nucleophilic substitution of the bromine atom may occur by the action of both carboxylate oxygen atom and C<sup>3</sup> of the heteroring. The latter pathway turned out to be preferred, and the products isolated after hydrolysis were 1-alkyl-1-aryl-2-oxo-1a,7b-dihydrocyclopropa-[c]chromene-1a-carboxylic acids XVa-XVe. In the IR spectra of compounds XVa, XVb, XVd, and XVe we observed absorption bands belonging to stretching vibrations of the aroyl, acid, and lactone carbonyl groups (1660-1715 and 1740-1745 cm<sup>-1</sup>) and O-H bond (3150–3180 cm<sup>-1</sup>). The 7b-H proton gave a signal at  $\delta$  3.52–3.82 ppm in the <sup>1</sup>H NMR spectra.

Double recrystallization of acids XVa, XVd, and XVe from methanol resulted in their partial or complete (compound XVb) esterification; for example, from acid XVb we obtained methyl 1-(4-chlorophenyl)-1-methyl-2-oxo-1a,7b-dihydrocyclopropa[c]chromene-1a-carboxylate (XVII). The structure of ester XVII was proved by independent synthesis via cyclopropanation of methyl 2-oxochromene-3-carboxylate (XVI) with zinc enolate IId (Scheme 4). According to our previous data, analogous reactions give exclusively stereoisomers with Z configuration of the arovl group and substituent on C<sup>1a</sup> [3], and the chemical shifts of the methyl protons range from  $\delta$  1.12 to 1.40 ppm. In the <sup>1</sup>H NMR spectra of compounds **XVa** and **XVb**, the 1-CH<sub>3</sub> signal appeared at  $\delta$  1.10– 1.27 ppm; therefore, we concluded that the examined reaction is strictly stereoselective and that the products are stereoisomers with Z configuration of the carboxy and aroyl groups. This configuration is favorable for intramolecular cyclization via nucleophilic addition of the acid hydroxy group at the aroyl carbonyl group to produce hydroxy lactone. For example, 1-(4-chlorophenyl)-1-methyl-2-oxo-1a,7b-dihydrocyclopropa[c]chromene-1a-carboxylic acid (XVb) could be converted into cyclic isomer XVIII.

To check the above assumption, we performed reactions of compounds **XVa**–**XVe** with carboxylic acid anhydrides; we hoped that the reaction will occur just with the hydroxy lactone isomer of **XVa**–**XVe**. In fact, by heating compounds **XVa**–**XVe** with acetic and propionic anhydrides we obtained 9c-alkyl-1-aryl-3,4-dioxo-9b,9c-dihydro-2,5-dioxacyclopenta[2,3]cyclopropa[1,2-a]naphthalen-1-yl acetates and propionates **XIXa**–**XIXe** (Scheme 5). The IR spectra of compounds **XIXa**–**XIXe** contained absorption bands at 1750 and 1805 cm<sup>-1</sup> due to lactone carbonyl groups in the six- and five-membered rings and ester carbonyl. The 9b-H signal appeared in the <sup>1</sup>H NMR spectra at δ 2.80–3.00 ppm.

#### Scheme 5.

 $Ar = 4-C1C_6H_4$ .

 $R = R' = Me, Ar = 4-MeC_6H_4$  (a),  $4-CIC_6H_4$  (b),  $4-BrC_6H_4$  (c);  $R = Me, R' = Et, Ar = 4-CIC_6H_4$  (d);  $R = Et, R' = Me, Ar = 4-CIC_6H_4$  (e).

### **EXPERIMENTAL**

The IR spectra were recorded on a UR-20 spectrometer from samples dispersed in mineral oil. The <sup>1</sup>H NMR spectra were measured from solutions in CDCl<sub>3</sub> (VIa, VIb, VId, VIe, XVa, XVII, XIXb,

XIXd, XIXe) or CDCl<sub>3</sub>–DMSO- $d_6$  (VIf, VIi, XVb, XIXa, XIXc) on a Tesla BS-576A instrument (100 MHz, HMDS) and from solutions in DMSO- $d_6$  (VIc–VIe, VIg, VIh, VIj, VIk, XVd, XVe) on a Bruker DRX-500 spectrometer (500 MHz, TMS). The <sup>13</sup>C NMR spectra of compounds VId, VIe, XI, and XII and their 1D and 2D NMR spectra were obtained from solutions in CDCl<sub>3</sub> on a Bruker DRX-400 spectrometer (400 and 100 MHz for <sup>1</sup>H and <sup>13</sup>C, respectively).

N-Substituted 2-alkyl-3-aryl-2-aroyl-1-cyanocyclopropane-1-carboxamides Va-Vk (general procedure). A solution of 0.025 mol of 1-aryl-2,2-dibromoalkanone I in 3 ml of ethyl acetate was added to a mixture of 4 g of zinc prepared as fine turnings, 8 ml of diethyl ether, and 5 ml of ethyl acetate. The mixture was heated to initiate the reaction which then occurred spontaneously. When the reaction was complete, the mixture was heated for 10 min under reflux, cooled, and separated from zinc by decanting. N-Substituted 3-aryl-2-cyanoprop-2-enamide, 0.01 mol, was added, the mixture was heated for 30 min, cooled, treated with a solution of acetic acid, and extracted with diethyl ether, the extract was dried over Na<sub>2</sub>SO<sub>4</sub>, the solvent was distilled off, and the residue was recrystallized from methanol.

*N*-Benzyl-1-cyano-2-(4-fluorobenzoyl)-2-methyl-3-phenylcyclopropane-1-carboxamide (VIa). Yield 66%, mp 159–160°C. IR spectrum, v, cm<sup>-1</sup>: 1680, 3390. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.50 s (3H, CH<sub>3</sub>), 3.54 s (1H, CH), 4.38 d (2H, CH<sub>2</sub>Ph, J = 7.1 Hz), 6.80–7.80 m (14H, H<sub>arom</sub>), 9.07 m (1H, NH). Found, %: C 75.59; H 5.08; N 6.69. C<sub>26</sub>H<sub>21</sub>FN<sub>2</sub>O<sub>2</sub>. Calculated, %: C 75.71; H 5.13; N 6.79.

*N*-Benzyl-1-cyano-2-ethyl-2-(4-fluorobenzoyl)-3-phenylcyclopropane-1-carboxamide (VIb). Yield 63%, mp 179–180°C. IR spectrum,  $\mathbf{v}$ , cm<sup>-1</sup>: 1680, 3395. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 0.65 t (3H, CH<sub>3</sub>CH<sub>2</sub>, J = 7.4 Hz), 1.89 q (2H, CH<sub>3</sub>CH<sub>2</sub>, J = 7.4 Hz), 3.58 s (1H, CH), 4.33 d (2H, CH<sub>2</sub>Ph, J = 7.1 Hz), 6.75–7.80 m (14H, H<sub>arom</sub>), 9.06 m (1H, NH). Found, %: C 75.96; H 5.35; N 6.45. C<sub>27</sub>H<sub>23</sub>FN<sub>2</sub>O<sub>2</sub>. Calculated, %: C 76.04; H 5.44; N 6.57.

**2-Benzoyl-3-(3-bromophenyl)-1-cyano-***N***-cyclohexyl-2-methylcyclopropane-1-carboxamide (VIc).** Yield 68%, mp 197–198°C. IR spectrum, v, cm<sup>-1</sup>: 1690, 3405.  $^{1}$ H NMR spectrum, δ, ppm: 1.54 s (3H, CH<sub>3</sub>), 1.10–1.80 m (10H, C<sub>6</sub>H<sub>11</sub>), 3.59 s (1H, CH), 3.66 m (1H, C<sub>6</sub>H<sub>11</sub>), 7.17–7.60 m and 7.87 d (9H, H<sub>arom</sub>), 8.56 m (1H, NH). Found, %: C 64.44; H 5.33;

N 5.93. C<sub>25</sub>H<sub>25</sub>BrN<sub>2</sub>O<sub>2</sub>. Calculated, %: C 64.52; H 5.41; N 6.02.

3-(3-Bromophenyl)-2-(4-chlorobenzoyl)-1-cyano-N-cyclohexyl-2-methylcyclopropane-1-carboxamide **(VId).** Yield 71%, mp 213–214°C. IR spectrum, v, cm<sup>-1</sup>: 1665, 3375. <sup>1</sup>H NMR spectrum, δ, ppm: 500 MHz, DMSO- $d_6$ : 1.52 s (3H, CH<sub>3</sub>); 1.05–1.80 m (10H,  $C_6H_{11}$ ); 3.58 s (1H, CH); 3.66 m (1H,  $C_6H_{11}$ ); 7.20 m, 7.41 m, 7.53 m, and 7.89 d (8H, 4-ClC<sub>6</sub>H<sub>4</sub>, 3-BrC<sub>6</sub>H<sub>4</sub>); 8.60 m (1H, NH); 400 MHz, CDCl<sub>3</sub>: 1.67 s (3H, CH<sub>3</sub>), 1.18-2.00 m (10H,  $C_6H_{11}$ ), 3.62 s (1H, 3-H), 3.84 m (1H, 1"-H), 6.56 d (1H, NH, J = 7.7 Hz), 6.95 d.t (1H, 6'-H, J = 7.9, 1.5 Hz), 7.30–7.34 m (3H, m-H, 4'-H), 7.46 t (1H, 2'-H, J = 1.8 Hz), 7.72 d (2H, o-H, J =8.6 Hz); 100 MHz, CDCl<sub>3</sub>: 1.58 s (3H, CH<sub>3</sub>), 1.15– 1.95 m (10H,  $C_6H_{11}$ ), 3.51 s (1H, CH), 3.72 m (1H,  $C_6H_{11}$ ), 6.42 d (1H, NH), 6.75–7.30 m and 7.55 d (8H, 4-ClC<sub>6</sub>H<sub>4</sub>, 3-BrC<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR spectrum, δ<sub>C</sub>, ppm: 18.40 q.d (CH<sub>3</sub>,  ${}^{1}J_{CH} = 131.2$ ,  ${}^{3}J_{CH} = 5.6$  Hz), 24.71 t.m (C<sup>3"</sup>,  ${}^{1}J_{CH} = 131.5$  Hz), 24.75 t.m (C<sup>5"</sup>,  ${}^{1}J_{CH} = 131.5$  Hz) 131.5 Hz), 25.31 t.m ( $C^{4"}$ ,  $^{1}J_{CH} = 126.0$ ), 30.97 d.q ( $C^{1}$ ,  $^{3}J_{CH} = ^{2}J_{CH} = 4.0$  Hz), 32.56 d.d.m ( $C^{6"}$ ,  $^{1}J_{CH} = 132.7$ , 128.0 Hz), 33.01 d.d.m ( $C^{2"}$ ,  $^{1}J_{CH} = 134.1$ , 130.0 Hz), 40.73 d.q ( $C^{3}$ ,  $^{1}J_{CH} = 163.7$ ,  $^{3}J_{CH} = 4.6$  Hz), 46.23 q.d  $(C^2, {}^2J_{C,Me} = 4.6, {}^2J_{C,3-H} = 2.7 \text{ Hz}), 50.20 \text{ d.m } (C^{1}),$  $^{1}J_{\text{CH}} = 143.2 \text{ Hz}$ ), 116.41 d (C=N,  $^{3}J_{\text{CH}} = 3.7 \text{ Hz}$ ), 122.52 d.t ( $C^{3'}$ , J = 8.1, 4.2 Hz), 125.78 d.d.t ( $C^{6'}$ , J =160.3, 7.3, 6.6 Hz), 129.12 d.d ( $C^m$ ,  $J_{CH} = 168.1$ , 5.1 Hz), 129.90 d ( $C^{5'}$ ,  ${}^{1}J_{CH} = 163.4$  Hz), 130.83 d.d  $(C^{o}, J_{CH} = 163.0, 6.7 \text{ Hz}), 131.22 \text{ d.d.d.d } (C^{4'}, J_{CH} =$ 167.4, 7.5, 5.3, 2.1 Hz), 131.82 t ( $C^{i}$ ,  ${}^{3}J_{CH} = 7.4$  Hz), 132.50 d.d.t ( $C^{2'}$ ,  $J_{CH} = 164.4$ , 6.1, 5.6 Hz), 134.23 d.m  $(C^{1}, J_{CH} = 8.0 \text{ Hz}), 140.56 \text{ t.t } (C^{p}, J_{CH} = 10.8, 3.3 \text{ Hz}),$ 160.92 t.d (NCO,  $J_{CH} = 4.2$ , 2.5 Hz), 193.03 m (CO). Found, %: C 59.98; H 4.79; N 5.49. C<sub>25</sub>H<sub>24</sub>ClBrN<sub>2</sub>O<sub>2</sub>. Calculated, %: C 60.08; H 4.84; N 5.60.

**3-(3-Bromophenyl)-2-(4-chlorobenzoyl)-1-cyano-***N*-cyclohexyl-2-ethylcyclopropane-1-carboxamide (VIe). Yield 64%, mp 199–200°C. IR spectrum, ν, cm<sup>-1</sup>: 1690, 3400. <sup>1</sup>H NMR spectrum, δ, ppm: 500 MHz, DMSO-*d*<sub>6</sub>: 0.82 t (3H, CH<sub>3</sub>, *J* = 7.4 Hz); 1.05–1.87 m (10H, C<sub>6</sub>H<sub>11</sub>, 1H, CH<sub>2</sub>CH<sub>3</sub>); 1.99 m (1H, CH<sub>2</sub>CH<sub>3</sub>); 3.64 s (1H, CH); 3.66 m (1H, C<sub>6</sub>H<sub>11</sub>); 7.20 m, 7.41 m, 7.51 m, and 7.93 d (8H, 4-ClC<sub>6</sub>H<sub>4</sub>, 3-BrC<sub>6</sub>H<sub>4</sub>); 8.14 m (1H, NH); 400 MHz, CDCl<sub>3</sub>: 0.83 t (3H, CH<sub>3</sub>, *J* = 7.4 Hz), 1.19–2.48 m (5H), 1.67 m (1H), 1.80 m (2H), 1.99 d (2H, CH<sub>2</sub>C<sub>6</sub>H<sub>11</sub>), 2.21 q (2H, CH<sub>2</sub>CH<sub>3</sub>, *J* = 7.4 Hz), 3.70 s (1H, 3-H), 3.87 m (1H, 1"-H), 6.63 d (1H, NH, *J* = 7.9 Hz), 6.98 d.t (1H, 6'-H, *J* = 7.9, 1.7 Hz), 7.03 t (1H, 5'-H, *J* = 7.9 Hz), 7.30–7.34 m (3H, *m*-H, 4'-H), 7.51 t (1H, 2'-H, *J* = 1.7 Hz),

7.74 d (2H, o-H, J = 8.7 Hz); 100 MHz, CDCl<sub>3</sub>: 0.74 t (3H, CH<sub>3</sub>, J = 7.4 Hz), 1.10–2.15 m (10H, C<sub>6</sub>H<sub>11</sub>, 2H,  $CH_2CH_3$ ), 3.59 s (1H, CH), 3.75 m (1H,  $C_6H_{11}$ ), 6.48 d (1H, NH, J = 7.9 Hz), 6.80–7.34 m and 7.56 d (8H, 4-ClC<sub>6</sub>H<sub>4</sub>, 3-BrC<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR spectrum,  $\delta_C$ , ppm: 10.68 q.t (CH<sub>3</sub>,  ${}^{1}J_{CH} = 126.7$ ,  ${}^{2}J_{CH} = 4.3$  Hz), 24.73 t.m  $(C^{3"}, C^{5"}, {}^{1}J_{CH} = 129.5 \text{ Hz}), 25.32 \text{ t.m } (C^{4"}, {}^{1}J_{CH} =$ 125 Hz), 25.48 t.m (2-CH<sub>2</sub>,  ${}^{1}J_{CH}$  = 131.2 Hz), 30.50 d.t  $(C^{1}, J_{CH} = 5.6, 2.9 \text{ Hz}), 32.55 \text{ t.m } (C^{6"}, {}^{1}J_{CH} = 130 \text{ Hz}), 32.67 \text{ t.m } (C^{2"}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m } (C^{3}, {}^{1}J_{CH} = 130 \text{ Hz}), 40.89 \text{ d.m} (C^{3}, {}^{1}J_{CH} = 130 \text{ d.m}), 40.89 \text{ d.m} (C^{3}, {}^{1}J_{CH} = 130 \text{ d.m}), 40.89 \text{ d.m} (C^{3}, {}^{1}J_{CH} = 130 \text{ d.m}), 40.89 \text{ d.m} (C^{3}, {}^{1}J_{CH} = 130$ 163.7 Hz), 50.12 d.m ( $C^{1}$ ,  ${}^{1}J_{CH} = 143.5$  Hz), 52.59 m  $(C^2)$ , 116.76 d  $(C \equiv N, {}^3J_{CH} = 4.0 \text{ Hz})$ , 122.56 d.d.d  $(C^3)$  $J_{\text{CH}} = 11.0, 4.5, 3.1 \text{ Hz}$ ), 125.75 d.d.t (C<sup>6</sup>,  $J_{\text{CH}} = 160.4$ , 6.9, 6.6 Hz), 129.09 d.d ( $C^m$ ,  $J_{CH} = 167.9$ , 5.2 Hz), 129.96 d ( $C^{5'}$ ,  ${}^{1}J_{CH} = 163.4$  Hz), 130.78 d.d ( $C^{o}$ ,  $J_{CH} = 163.4$  Hz), 130.78 d.d ( $C^{o}$ ),  $J_{CH} = 163.4$  Hz) 163.2, 6.9 Hz), 131.26 d.d.d.d ( $C^4$ ,  $J_{CH} = 167.7$ , 7.1, 5.1, 1.9 Hz), 132.12 t (C',  ${}^{3}J_{CH} = 7.4$  Hz), 132.61 d.q  $(C^2, J_{CH} = 164.5, 6.0 \text{ Hz}), 134.32 \text{ d.q } (C^1, J_{CH} = 8.5,$ 1.3 Hz), 140.48 t.t ( $C^p$ ,  $J_{CH} = 10.7$ , 3.3 Hz), 161.10 t.d (NCO,  $J_{CH}$  = 4.4, 2.4 Hz), 192.42 m (CO). Found, %: C 60.71; H 5.03; N 5.32. C<sub>26</sub>H<sub>26</sub>ClBrN<sub>2</sub>O<sub>2</sub>. Calculated, %: C 60.77; H 5.10; N 5.45.

978

**2-Benzoyl-1-cyano-2-methyl-***N***-(3-methylphenyl)-3-phenylcyclopropane-1-carboxamide (VIf).** Yield 67%, mp 200–201°C. IR spectrum, v, cm<sup>-1</sup>: 1690, 3325.  $^{1}$ H NMR spectrum,  $\delta$ , ppm: 1.59 s (3H, CH<sub>3</sub>), 2.24 s (3H, CH<sub>3</sub>), 3.61 s (1H, CH), 6.80–7.68 m (14H, H<sub>arom</sub>), 10.22 s (1H, NH). Found, %: C 79.06; H 5.53; N 6.94. C<sub>26</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>. Calculated, %: C 79.17; H 5.62; N 7.10.

**2-(4-Chlorobenzoyl)-1-cyano-2-methyl-***N***-(3-methylphenyl)-3-phenylcyclopropane-1-carbox-amide (VIg).** Yield 72%, mp 222–223°C. IR spectrum, ν, cm<sup>-1</sup>: 1675, 3330. <sup>1</sup>H NMR spectrum, δ, ppm: 1.59 s (3H, CH<sub>3</sub>), 2.30 s (3H, CH<sub>3</sub>), 3.71 s (1H, CH), 6.98–7.86 m (13H, H<sub>arom</sub>), 10.71 s (1H, NH). Found, %: C 72.75; H 4.87; N 6.39. C<sub>26</sub>H<sub>22</sub>ClN<sub>2</sub>O<sub>2</sub>. Calculated, %: C 72.81; H 4.93; N 6.53.

**2-Benzoyl-1-cyano-2-ethyl-***N***-(3-methylphenyl)-3-phenylcyclopropane-1-carboxamide (VIh).** Yield 61%, mp 227–228°C. IR spectrum, v, cm<sup>-1</sup>: 1685, 3330.  $^{1}$ H NMR spectrum, δ, ppm: 0.82 t (3H, CH<sub>3</sub>, J = 7.4 Hz), 1.79 m and 2.18 m (2H, CH<sub>2</sub>), 3.80 s (1H, CH), 6.98–7.88 m (14H, H<sub>arom</sub>), 10.72 s (1H, NH). Found, %: C 79.27; H 5.88; N 6.70. C<sub>27</sub>H<sub>24</sub>N<sub>2</sub>O<sub>2</sub>. Calculated, %: C 79.39; H 5.92; N 6.86

**2-(4-Chlorobenzoyl)-1-cyano-2-ethyl-***N***-(3-methylphenyl)-3-phenylcyclopropane-1-carboxamide** (VIi). Yield 58%, mp 252–253°C. IR spectrum, v, cm<sup>-1</sup>: 1685, 3330. <sup>1</sup>H NMR spectrum, δ, ppm: 0.81 t

(3H, CH<sub>3</sub>, J = 7.4 Hz), 1.95 m (2H, CH<sub>2</sub>), 3.66 s (1H, CH), 6.80–7.69 m (13H, H<sub>arom</sub>), 10.32 s (1H, NH). Found, %: C 73.10; H 5.17; N 6.21. C<sub>27</sub>H<sub>23</sub>ClN<sub>2</sub>O<sub>2</sub>. Calculated, %: C 73.21; H 5.23; N 6.32

**2-Benzoyl-1-cyano-***N***-(2-methoxyphenyl)-2-methyl-3-phenylcyclopropane-1-carboxamide (VIj).** Yield 42%, mp 177–179°C. IR spectrum, v, cm<sup>-1</sup>: 1680, 3395.  $^{1}$ H NMR spectrum, δ, ppm: 1.67 s (3H, CH<sub>3</sub>), 3.75 s (1H, CH), 3.86 s (3H, OCH<sub>3</sub>), 6.98–7.87 m (14H, H<sub>arom</sub>), 9.79 s (1H, NH). Found, %: C 76.01; H 5.32; N 6.69.  $C_{26}H_{22}N_2O_3$ . Calculated, %: C 76.08; H 5.40; N 6.82.

**2-(4-Chlorobenzoyl)-1-cyano-***N***-(2-methoxyphenyl)-2-methyl-3-phenylcyclopropane-1-carboxamide (VIk).** Yield 38%, mp 173–174°C. IR spectrum,  $\nu$ , cm<sup>-1</sup>: 1695, 3395. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.66 s (3H, CH<sub>3</sub>), 3.74 s (1H, CH), 3.84 s (3H, OCH<sub>3</sub>), 6.98–7.88 m (13H, H<sub>arom</sub>), 9.82 s (1H, NH). Found, %: C 70.08; H 4.70; N 6.18. C<sub>26</sub>H<sub>21</sub>ClN<sub>2</sub>O<sub>3</sub>. Calculated, %: C 70.19; H 4.76; N 6.30.

6-(4-Bromobenzoyl)-4,4,6-trimethyl-2-oxo-3-oxabicyclo[3.1.0]hexane-1-carboxylic acid ethyl ester and morpholide XI and XII (general procedure). A solution of 0.013 mol of 2,2-dibromo-1-(4-bromophenyl)propan-1-one in 3 ml of ethyl acetate was added to a mixture of 3 g of zinc prepared as fine turnings, 8 ml of diethyl ether, and 5 ml of ethyl acetate. The mixture was heated to initiate the reaction which then occurred spontaneously. When the reaction was complete, the mixture was heated for 10 min under reflux, cooled, and separated from zinc by decanting. 5,5-Dimethyl-2-oxo-2,5-dihydrofuran-3carboxylic acid ethyl ester or morpholide, 0.01 mol, was added, the mixture was heated for 30 min, cooled, treated with a solution of acetic acid, and extracted with diethyl ether, the extract was dried over Na<sub>2</sub>SO<sub>4</sub>, the solvent was distilled off, and the residue was recrystallized from methanol.

Ethyl 6-(4-bromobenzoyl)-4,4,6-trimethyl-2-oxo-3-oxabicyclo[3.1.0]hexane-1-carboxylate (XI). Yield 73%, mp 160–161°C. IR spectrum, v, cm<sup>-1</sup>: 1680, 1720, 1770. <sup>1</sup>H NMR spectrum, δ, ppm: 1.39 t (3H, CH<sub>3</sub>CH<sub>2</sub>, J = 7.1 Hz), 1.44 s (3H, 4-CH<sub>3</sub>), 1.51 s (3H, 4-CH<sub>3</sub>), 1.59 s (3H, 6-CH<sub>3</sub>), 2.59 s (1H, CH), 4.38 q (2H, CH<sub>3</sub>CH<sub>2</sub>, J = 7.1 Hz), 7.64 d and 7.98 d (4H, BrC<sub>6</sub>H<sub>4</sub>, J = 8.7 Hz). <sup>13</sup>C NMR spectrum, δ<sub>C</sub>, ppm: 14.20 q.t (CH<sub>2</sub>CH<sub>3</sub>, J = 127.4, 2.6 Hz), 20.15 q.d (6-CH<sub>3</sub>, J = 130.9, 5.2 Hz), 22.21 q.q (4-CH<sub>3</sub>, J = 128.1, 3.8 Hz), 31.07 q.q.d (4-CH<sub>3</sub>, J = 127.8, 4.2, 2.3 Hz), 42.61 d.q (C<sup>1</sup> or C<sup>6</sup>, J = 3.7 Hz), 42.84 q (C<sup>6</sup>

or C<sup>1</sup>, J = 4.8 Hz), 44.76 d.m (C<sup>5</sup>, J = 168.2 Hz), 62.49 t.q (C**H**<sub>2</sub>CH<sub>3</sub>, J = 148.5, 4.4 Hz), 82.40 m (C<sup>4</sup>), 129.47 t.t (C<sup>p</sup>, J = 10.9, 3.1 Hz), 131.11 d.d (C<sup>o</sup>, J = 163.9, 6.9 Hz), 132.01 d.d (C<sup>m</sup>, J = 168.4, 5.4 Hz), 133.12 t (C<sup>i</sup>, J = 7.4 Hz), 164.60 t.d (COO, J = 3.4 Hz), 168.09 d (C<sup>2</sup>, J = 4.0 Hz), 194.32 m (CO). Found, %: C 54.56; H 4.74. C<sub>18</sub>H<sub>19</sub>BrO<sub>5</sub>. Calculated, %: C 54.70; H 4.85.

6-(4-Bromobenzoyl)-1-morpholinocarbonyl-4,4,6-trimethyl-3-oxabicyclo[3.1.0]hexan-2-one (XII). Yield 64%, mp 237–238°C. IR spectrum, v, cm<sup>-1</sup>: 1665, 1750. <sup>1</sup>H NMR spectrum, δ, ppm: 1.19 s (3H, 4-CH<sub>3</sub>), 1.44 s (3H, 6-CH<sub>3</sub>), 1.57 s (3H, 4-CH<sub>3</sub>), 2.80 s (1H, 5-H), 3.38-4.39 m (8H, NCH<sub>2</sub>CH<sub>2</sub>O), 7.68 d and 7.89 d (4H,  $BrC_6H_4$ , J = 8.7 Hz). <sup>13</sup>C NMR spectrum,  $\delta_{\rm C}$ , ppm: 19.85 q.d (6-CH<sub>3</sub>, J = 130.1, 5.4 Hz), 24.88 q.q (4-CH<sub>3</sub>, J = 127.7, 3.8 Hz), 30.60 q.q.d (4-CH<sub>3</sub>, J = 127.9, 3.8, 3.6 Hz), 42.91 m  $(C^1 \text{ or } C^6)$ , 42.91 t (NCH<sub>2</sub>, J = 139.6 Hz), 43.61 q ( $C^6$ or  $C^1$ , J = 4.5 Hz), 46.91 d.q.q ( $C^5$ , J = 169.8, 4.2 Hz), 46.96 t (NCH<sub>2</sub>, J = 140.6 Hz), 66.60 t and 67.55 t  $(OCH_2, J = 144 Hz), 81.89 q.q.d (C^4, J = 4.4 Hz),$ 129.46 t.t ( $C^p$ , J = 10.9, 3.1 Hz), 131.31 d.d.d ( $C^o$ , J =164.4, 7.1, 4.4 Hz), 132.16 d.d.d ( $C^m$ , J = 166.6, 8.0, 3.4 Hz), 133.42 t ( $C^{i}$ , J = 7.4 Hz), 161.82 d.t (NCO, J = 2.8 Hz), 169.46 d (C<sup>2</sup>, J = 3.7 Hz), 194.59 m (CO). Found, %: C 54.94; H 4.96; N 3.05. C<sub>20</sub>H<sub>22</sub>BrNO<sub>5</sub>. Calculated, %: C 55.06; H 5.08; N 3.21.

1-Alkvl-1-arvl-2-oxo-1a,7b-dihvdrocvclopropa-[c]chromene-1a-carboxylic acids XVa-XVe (general procedure). A solution of 0.025 mol of 1-aryl-2,2dibromoalkanone **Ib** or **Id**–**Ig** in 10 ml of ethyl acetate was added dropwise under stirring to a mixture of 5 g of zinc prepared as fine turnings, 10 ml of diethyl ether, and 10 ml of ethyl acetate. The mixture was heated to initiate the reaction which then occurred spontaneously. When the reaction was complete, the mixture was heated for 15 min on a water bath, cooled, and separated from zinc by decanting into another flask. Compound XIII, 0.01 mol, was then added, the mixture was heated for 30 min under reflux, cooled. treated with 5% hydrochloric acid, and extracted with diethyl ether, the extract was dried over Na<sub>2</sub>SO<sub>4</sub>, the solvent was distilled off, and the residue was recrystallized from benzene-hexane.

1-Methyl-2-oxo-1-(p-tolyl)-1a,7b-dihydrocyclo-propa[c]chromene-1a-carboxylic acid (XVa). Yield 60%, mp 205–207°C. IR spectrum, v, cm<sup>-1</sup>: 1660, 1715, 1740, 3170. <sup>1</sup>H NMR spectrum, δ, ppm: 1.27 s (3H, CH<sub>3</sub>), 2.32 s (3H, CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>), 3.82 s (1H, CH), 6.85–7.80 m (8H, C<sub>6</sub>H<sub>4</sub>, CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>), 10.85 br.s (1H,

OH). Found, %: C 71.30; H 4.71. C<sub>20</sub>H<sub>16</sub>O<sub>5</sub>. Calculated, %: C 71.42; H 4.80.

**1-(4-Chlorophenyl)-1-methyl-2-oxo-1a,7b-dihydrocyclopropa**[*c*]**chromene-1a-carboxylic acid** (**XVb).** Yield 52%, mp 218–220°C. IR spectrum, v, cm<sup>-1</sup>: 1670–1700, 1740, 3150.  $^{1}$ H NMR spectrum, δ, ppm: 1.11 s (3H, CH<sub>3</sub>), 3.52 s (1H, CH), 6.85–7.85 m (8H, C<sub>6</sub>H<sub>4</sub>, 4-ClC<sub>6</sub>H<sub>4</sub>), 13.10 br.s (1H, OH). Found, %: C 63.91; H 3.61. C<sub>19</sub>H<sub>13</sub>ClO<sub>5</sub>. Calculated, %: C 63.97; H 3.67.

1-(4-Bromophenyl)-1-methyl-2-oxo-1a,7b-dihydrocyclopropa[c]chromene-1a-carboxylic acid (XVc) was brought into reaction with acetic anhydride without preliminary purification.

**1-Ethyl-2-oxo-1-phenyl-1a,7b-dihydrocyclo-propa**[*c*]**chromene-1a-carboxylic acid (XVd).** Yield 55%, mp 193–195°C. IR spectrum, v, cm<sup>-1</sup>: 1670–1700, 1745, 3180. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 0.46 t (3H, CH<sub>3</sub>CH<sub>2</sub>, J=7 Hz), 1.06 m and 1.98 m (2H, CH<sub>3</sub>CH<sub>2</sub>, J=7 Hz), 3.62 s (1H, CH), 7.15–8.00 m (9H, C<sub>6</sub>H<sub>4</sub>, C<sub>6</sub>H<sub>5</sub>), 13.45 br.s (1H, OH). Found, %: C 71.48; H 4.82. C<sub>20</sub>H<sub>16</sub>O<sub>5</sub>. Calculated, %: C 71.42; H 4.80.

**1-Ethyl-2-oxo-1-**(*p***-tolyl**)**-1a,7b-dihydrocyclo-propa**[*c*]**chromene-1a-carboxylic acid (XVe).** Yield 56%, mp 234–236°C. IR spectrum, v, cm<sup>-1</sup>: 1670–1700, 1740, 3170. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 0.46 t (3H, CH<sub>3</sub>CH<sub>2</sub>, J = 7 Hz), 1.05 m and 1.95 m (2H, CH<sub>3</sub>CH<sub>2</sub>, J = 7 Hz), 2.39 s (3H, 4-CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>), 3.61 s (1H, CH), 7.15–7.86 m (8H, C<sub>6</sub>H<sub>4</sub>, 4-CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>), 13.35 br.s (1H, OH). Found, %: C 72.09; H 5.25. C<sub>21</sub>H<sub>18</sub>O<sub>5</sub>. Calculated, %: C 71.99; H 5.18.

Methyl 1-(4-chlorophenyl)-1-methyl-2-oxo-1a,7b-dihydrocyclopropa[c]chromene-1a-carboxylate (XVII). a. A solution of 0.015 mol of 2,2-dibromo-1-(4-chlorophenyl)propan-1-one in 10 ml of ethyl acetate was added dropwise under stirring to a mixture of 5 g of zinc prepared as fine turnings, 10 ml of diethyl ether, and 10 ml of ethyl acetate. The mixture was heated to initiate the reaction which then occurred spontaneously. When the reaction was complete, the mixture was heated for 15 min on a water bath, cooled, and separated from zinc by decanting into another flask. Compound XVI, 0.01 mol, was then added, the mixture was heated for 30 min under reflux. cooled, treated with 5% hydrochloric acid, and extracted with diethyl ether, the extract was dried over Na<sub>2</sub>SO<sub>4</sub>, the solvent was distilled off, and the residue was recrystallized from benzene-hexane. Yield 82%.

b. Acid **XVb** was heated for 1 h in boiling methanol. The solution was cooled to -5°C, and compound

**XVII** was filtered off and dried under reduced pressure. Yield 61%, mp 216–218°C. IR spectrum,  $\nu$ , cm<sup>-1</sup>: 1690, 1735, 1770. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.17 s (3H, CH<sub>3</sub>), 2.49 s (3H, OCH<sub>3</sub>), 3.60 s (1H, CH), 6.90–7.93 m (8H, C<sub>6</sub>H<sub>4</sub>, ClC<sub>6</sub>H<sub>4</sub>). Found, %: C 64.91; H 4.15. C<sub>20</sub>H<sub>16</sub>O<sub>5</sub>. Calculated, %: C 64.79; H 4.08.

1-Aryl-9c-methyl-3,4-dioxo-9b,9c-dihydro-2,5-dioxacyclopenta[2,3]cyclopropa[1,2-a]naphthalen-1-yl acetates and propionate XIXa-XIXe (general procedure). Compound XVa-XVe, 0.005 mol, was added to 5 ml of acetic or propionic anhydride, and the mixture was heated for 1 h. Excess anhydride was distilled off under reduced pressure, and the residue was recrystallized from toluene.

9c-Methyl-3,4-dioxo-1-(p-tolyl)-9b,9c-dihydro-2,5-dioxacyclopenta[2,3]cyclopropa[1,2-a]naphthalen-1-yl acetate (XIXa). Yield 50%, mp 221–223°C. IR spectrum, v, cm<sup>-1</sup>: 1750, 1805. <sup>1</sup>H NMR spectrum, δ, ppm: 1.18 s (3H, CH<sub>3</sub>), 2.08 s (3H, OCOCH<sub>3</sub>), 2.30 s (3H, 4-CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>), 3.00 s (1H, CH), 6.80–7.30 m (8H, C<sub>6</sub>H<sub>4</sub>, 4-CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>). Found, %: C 69.72; H 4.73. C<sub>22</sub>H<sub>18</sub>O<sub>6</sub>. Calculated, %: C 69.84; H 4.80.

1-(4-Chlorophenyl)-9c-methyl-3,4-dioxo-9b,9c-dihydro-2,5-dioxacyclopenta[2,3]cyclopropa[1,2-a]-naphthalen-1-yl acetate (XIXb). Yield 47%, mp 223–224°C. IR spectrum, v, cm<sup>-1</sup>: 1750, 1805. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.22 s (3H, CH<sub>3</sub>), 2.09 s (3H, OCOCH<sub>3</sub>), 2.80 s (1H, CH), 6.80–7.30 m (8H, C<sub>6</sub>H<sub>4</sub>, 4-ClC<sub>6</sub>H<sub>4</sub>). Found, %: C 63.30; H 3.86. C<sub>21</sub>H<sub>15</sub>ClO<sub>6</sub>. Calculated, %: C 63.25; H 3.80.

1-(4-Bromophenyl)-9c-methyl-3,4-dioxo-9b,9c-dihydro-2,5-dioxacyclopenta[2,3]cyclopropa[1,2-a]-naphthalen-1-yl acetate (XIXc). Yield 58%, mp 240–242°C. IR spectrum, v, cm<sup>-1</sup>: 1750, 1805. <sup>1</sup>H NMR spectrum, δ, ppm: 1.20 s (3H, CH<sub>3</sub>), 2.10 s (3H, OCOCH<sub>3</sub>), 3.06 s (1H, CH), 6.80–7.40 m (8H, C<sub>6</sub>H<sub>4</sub>, 4-BrC<sub>6</sub>H<sub>4</sub>). Found, %: C 57.01; H 3.48. C<sub>21</sub>H<sub>15</sub>BrO<sub>6</sub>. Calculated, %: C 56.90; H 3.41.

1-(4-Chlorophenyl)-9c-methyl-3,4-dioxo-9b,9c-dihydro-2,5-dioxacyclopenta[2,3]cyclopropa[1,2-a]-naphthalen-1-yl propionate (XIXd). Yield 52%, mp 240–241°C. IR spectrum, v, cm<sup>-1</sup>: 1750, 1805. <sup>1</sup>H NMR spectrum, δ, ppm: 1.10 t (3H, CH<sub>3</sub>CH<sub>2</sub>CO, J = 7 Hz), 1.20 s (3H, CH<sub>3</sub>), 2.35 q (2H, CH<sub>3</sub>CH<sub>2</sub>CO, J = 7 Hz), 2.80 s (1H, CH), 6.75–7.30 m (8H, C<sub>6</sub>H<sub>4</sub>, 4-ClC<sub>6</sub>H<sub>4</sub>). Found, %: C 63.50; H 4.07. C<sub>22</sub>H<sub>17</sub>ClO<sub>6</sub>. Calculated, %: C 64.01; H 4.15.

1-(4-Chlorophenyl)-9c-ethyl-3,4-dioxo-9b,9c-di-hydro-2,5-dioxacyclopenta[2,3]cyclopropa[1,2-a]-naphthalen-1-yl acetate (XIXe). Yield 64%, mp 189–191°C. IR spectrum, v, cm<sup>-1</sup>: 1750, 1805. <sup>1</sup>H NMR spectrum, δ, ppm: 0.80 t (3H, CH<sub>3</sub>, J = 7 Hz), 1.65 q (2H, CH<sub>3</sub>CH<sub>2</sub>, J = 7 Hz), 2.10 s (3H, OCOCH<sub>3</sub>), 2.95 s (1H, CH), 6.75–7.30 m (8H, C<sub>6</sub>H<sub>4</sub>, 4-ClC<sub>6</sub>H<sub>4</sub>). Found, %: C 64.10; H 3.89. C<sub>22</sub>H<sub>17</sub>ClO<sub>6</sub>. Calculated, %: C 64.01; H 4.15.

This study was performed under financial support by the Russian Foundation for Basic Research (project nos. 04-03-96036, 04-03-97505) and by the Federal Education Agency (project no. A.04-2.11-492).

#### REFERENCES

- 1. Aliev, Z.G., Shchepin, V.V., Lewis, S.B., Shchepin, R.V., and Atovmyan, L.O., *Izv. Ross. Akad. Nauk, Ser. Khim.*, 2000, p. 2107.
- 2. Shchepin, V.V., Tryastsin, A.A., Shchepin, R.V., Kalyuzhnyi, M.M., and Lewis, Scott B., *Russ. J. Org. Chem.*, 2001, vol. 37, p. 1596.
- 3. Shchepin, V.V., Kalyuzhnyi, M.M., Silaichev, P.S., Russkikh, N.Yu., Shchepin, R.V., Ezhikova, M.A., and Kodess, M.I., *Russ. J. Org. Chem.*, 2004, vol. 40, p. 1353.
- 4. Shchepin, V.V., Silaichev, P.S., Shchepin, R.V., Ezhikova, M.A., and Kodess, M.I., *Russ. J. Org. Chem.*, 2005, vol. 41, p. 527.
- 5. Shchepin, V.V., Kalyuzhnyi, M.M., Shchepin, R.V., and Vakhrin, M.I., *Russ. J. Gen. Chem.*, 2003, vol. 73, p. 758.